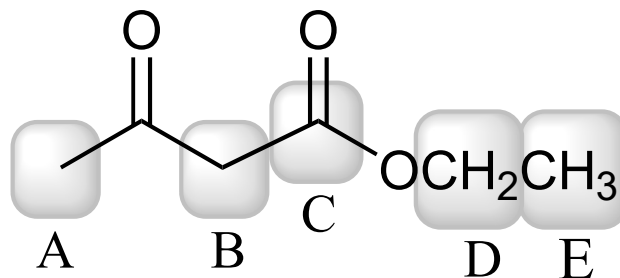


Identify the most acidic proton(s) in the following compound.



Provide an explanation for your answer.

- I. The conj. base is stabilized by inductive effects.
- II. The negative charge in conj. base is on an electronegative atom.
- III. The conj. base is stabilized by a large amount of resonance.

- A) I only
- B) II only
- C) III only
- D) Both I and II
- E) Both II and III